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## Ultimate Chemical Equations Handbook Answers Chapter 9

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Answers for the questions in Chapter 8: ROUND 3 Exercise 8-1: Using the activity series, predict and balance the following single replacement reactions. Use abbreviations to indicate the appropriate phase of reactants and products where possible. Note: Not all of the reactions will occur. For those that do not, write no reaction. I.

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Ultimate Chemical Equations Handbook Answers Answers for the questions in Chapter 8: ROUND 3 Exercise 8-1: Using the activity series, predict and balance the following single replacement reactions. Use abbreviations to indicate the

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appropriate phase of reactants and products where possible.

## Ultimate Chemical Equations Handbook Answers Chapter 10

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The Ultimate Chemical Equations Handbook Determine the oxidation number of the underlined element:  $\text{KMnO}_4$  Since K is an alkali metal, its charge must be  $1+$  Oxygen is  $2-$  but there are four of them, therefore,  $4 \times 2-$  equals  $8-$  If  $1+$  and  $8-$  are added Ultimate Chemical Equation Answers

## The Ultimate Chemical Equations Handbook Answers Chapter 7

The Ultimate Chemical Equations Handbook Since many of the transuranium elements were prepared at the Lawrence Radiation Center, University of California at Berkeley, California, U.S.A., it is logical to find the following names: Americium—named after the Americas (after all, one must be patriotic!).

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